

Editorial: Proceedings of the 2015 Workshop on the Determination of the Fundamental Constants

Cite as: J. Phys. Chem. Ref. Data **44**, 030401 (2015); <https://doi.org/10.1063/1.4926441>
Submitted: 25 June 2015 . Accepted: 25 June 2015 . Published Online: 17 July 2015

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Editorial: Proceedings of the 2015 Workshop on the Determination of the Fundamental Constants

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(Received 25 June 2015; accepted 25 June 2015; published online 17 July 2015)

[<http://dx.doi.org/10.1063/1.4926441>]

CONTENTS

1. Overview of the Workshop and Proceedings	1
2. Overview of the Fundamental Constants	1
3. History of Evaluations of the Fundamental Constants	2
4. References	3

1. Overview of the Workshop and Proceedings

The *Workshop on the Determination of the Fundamental Constants* was held on February 1–6, 2015, at the Hotel Frankenbach in Eltville, Germany.¹ The workshop involved approximately 70 presentations (talks and posters); the 15 papers submitted as part of the proceedings covered a wide range of fundamental constants, including the Rydberg constant R_{∞} , fine-structure constant α , Avogadro constant N_A , proton charge radius r_p , bound-electron g factor, heliocentric gravitational constant GM_{Sun} , anomalous magnetic moment of the muon a_{μ} , hyperfine splitting of positronium, root mean square (rms) nuclear charge radii $R = \langle r^2 \rangle^{1/2}$, and nuclear magnetic dipole μ and electric quadrupole Q moments of various nuclei. No papers were submitted for two of the most basic fundamental constants: the Boltzmann constant k_B and the Planck constant h . [Note: We have included the program for the workshop as the supplementary material⁹¹].

The proceedings can be considered as supporting information for the report “Committee on Data for Science and Technology (CODATA) Recommended Values of the Fundamental Physical Constants: 2014,” which will be published simultaneously in the *Journal of Physical Chemical Reference Data* and the *Review of Modern Physics* in 2016. The report—which updates the earlier CODATA 2010 recommendations^{2,3}—is a product of the *CODATA Task Group on Fundamental Constants*.⁴

Similar to the previous reports, the 2014 CODATA report will summarize the available data related to each of the fundamental constants and provide an overview of the new experimental techniques and computational methods used to produce the data that determine values for the fundamental

constants. The report provides an analysis of the available data, evaluates an uncertainty for each datum, and when necessary determines weighting factors to account for possible unknown systematic uncertainties. It also considers and evaluates correlations between fundamental constants that are interdependent and coupled. Because of the volume of data considered in the fundamental constants evaluation, the report provides only a brief overview about the background and motivation of the various experimental and computational efforts. In addition, the report only briefly summarizes the physical theory, numerical calculations, uncertainty analysis, and other factors pertaining to each measurement or calculation.

Each article in these proceedings provides supporting information to the fundamental constants report. These articles provide much more detail regarding uncertainty analysis and other considerations needed to produce a self-consistent set of best values. A more extensive overview of the proceedings and presentations at the Workshop can be found in an Introductory Paper in this issue entitled “Advances in determination of fundamental constants” by Karshenboim, Mohr, and Newell.⁵

2. Overview of the Fundamental Constants

The fundamental constants of physics and chemistry are an internationally accepted, self-consistent set of best values (and their uncertainties) used by scientists and engineers in quantifying and reporting measurements and calculations of physical phenomena. Accurate fundamental constants are necessary to compare different types of related observations. Many of the fundamental constants are basic constants of proportionality between units of measurement and physical quantities. For example, the Boltzmann constant k_B provides the relationship between temperature and the energy of a particle through $E = k_B T$, while Planck’s constant h provides the relationship between the frequency of light and the energy of a photon through $E = h\nu$. A number of other fundamental constants, however, characterize complicated interactions. For example, the fine-structure constant α characterizes the electromagnetic interactions of charged particles and is related to other constants by $\alpha = e^2/2\varepsilon_0 hc$, where e , ε_0 , and c are the elementary charge, electric constant (vacuum permittivity), and speed of light, respectively.^{2,3}

Measurement technology has advanced to a state where many of the fundamental constants are known to great accuracy.^{2,3} For example, the meter, the unit of length, is now defined^{6,7} in terms of an exact value for the speed of light c in the SI system of units ($c = 299\,792\,458$ m/s).^{8,9} The Rydberg

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constant R_∞ and the fine-structure constant α have very small relative standard uncertainties of about 5×10^{-12} and 3×10^{-10} , respectively. Many of the fundamental constants^{2,3}—including the Planck constant h , elementary charge e , electron mass m_e , proton mass m_p , and Avogadro's number N_A —have somewhat higher, but still small, relative standard uncertainties of about $(2\text{--}4) \times 10^{-8}$. The relative standard uncertainty of the Boltzmann constant k_B is substantially higher at about 1×10^{-6} .

There are two fundamental constants that each have significant unresolved discrepancies, where different types of measurements produce values that differ by many times their standard deviations.^{2,3} The accepted value of the Newtonian constant of gravitation G has a high relative standard uncertainty of about 1.2×10^{-4} , because several measurements differ by about $(10\text{--}15)\sigma$ of their individual standard uncertainties. Similarly, the proton rms charge radius r_p has a high relative standard uncertainty of about 6×10^{-3} with two types of measurements differing by about $(5\text{--}7)\sigma$ of their individual standard uncertainties. This tension in the r_p values is termed the “proton radius puzzle.” In both cases, G and r_p , the question remains: are there unidentified systematic errors in some of the measurements, or is the physics of these phenomena not sufficiently understood?

3. History of Evaluations of the Fundamental Constants

Compilations and evaluations of the fundamental constants date to the 1920s. Overviews of the history of this effort have been given previously by Rossini (Chair of the International Union of Pure and Applied Chemistry (IUPAC) Task Group on Fundamental Constants) in 1964¹⁰ and in a report by the U.S. National Research Council (NRC) Committee of Fundamental Constants in 1983.¹¹ We update and expand on this history here.

In 1919, the IUPAC approved a project called the “International Critical Tables (ICT),” which was to provide recommended values for a wide range of properties in physics, chemistry, and technology. The ICT Advisory Committees were established several years later in 1922. The values for the fundamental constants in the ICT were prepared under the direction of Frederick E. Fowle of the Smithsonian Institution based on information obtained from various laboratories throughout the world. These values were adopted in 1923 and then published as a table in the first volume of the “International Critical Tables of Numerical Data, Physics, Chemistry and Technology” in 1926.¹² No references, evaluation, or discussion were provided for the recommended values.

In 1926, Henning and Jaeger¹³ published a chapter in the “Handbuch der Physik” entitled “The General Physical Constants” that provided some discussion of the selected values and references. Also in 1926, Birge published an article in *Science* entitled “The most probable value of certain basic constants” discussing the recommended values from the *International Critical Tables*, pointing out inconsistencies,

sources of error, and the interdependencies of many of the fundamental constants.¹⁴

In 1928, the U.S. National Research Council Division of Physical Sciences established the “Committee on Physical Constants.” The stated purpose of the committee was “to encourage the preparation of papers on the general physical constants and derived constants whenever a significant change amounting to more than the probable error of a new value is reported.” Raymond Birge was the chair of the committee and largely responsible for its output—working for the most part independently.

In 1929, the first comprehensive review of the fundamental constants was published by Birge as the first article in the first volume of *Reviews of Modern Physics*.¹⁵ (About a dozen fundamental constants’ updates have been published in *Reviews of Modern Physics*). In 1932, Birge reevaluated a number of the fundamental constants using a self-consistent least squares analysis¹⁶ but found little statistical differences with his 1929 initial evaluation. The next major update to the fundamental constants by Birge was published in two papers in 1941.^{17,18} He also published an extensive set of papers regarding evaluations of the fundamental constants in general^{19–22} and for specific subsets of fundamental constants.^{23–33}

During the 1930–1940s, a number of authors published compilations and evaluations of various subsets of the fundamental constants, including Born (1930),³⁴ Millikan (1930,1938),^{35,36} Bond (1936),³⁷ von Friesen (1937),³⁸ Dunington (1939),³⁹ Wensel,⁴⁰ and Stille (1943 and 1948).^{41,42}

In 1939, DuMond published a paper in *Physical Review* discussing the interdependencies of the fundamental constants.⁴³ The following year, the National Research Council established the “Committee of Fundamental Constants and Conversion Factors” for the “purpose of ascertaining and publishing from time to time the best and most generally acceptable values of the physical constants.”

Working under the auspices of the U.S. NRC Subcommittee on Fundamental Constants, Dumond and Cohen produced major updates to the fundamental constants (and associated intermediate discussion papers) in 1947,^{44–47} 1950,^{48,49} 1952,^{50,51} 1955,^{52–55} 1959,⁵⁶ and 1965.^{57,58} During this time period, Bearden and coworkers also contributed to the work of this NRC subcommittee and published a series of papers making recommendations for some of the fundamental constants and providing some discussion about inconsistencies, uncertainties, and evaluation methods.^{59–64} A summary report by the NRC committee was published in 1952.⁶⁵

In 1966, the CODATA was established by the International Council of Science, and in 1969, the CODATA Task Group on Fundamental Constants was formed. Also in 1969, Taylor, Parker, and Langenberg published an update to the fundamental constants⁶⁶ based upon a new determination of e/h by Taylor *et al.* in 1967.⁶⁷ In 1973, Cohen and Taylor produced the first CODATA recommended values for the fundamental constants⁶⁸ followed by an update in 1986.^{69–72} Mohr and Taylor at the National Institute of Standards and Technology (NIST) revised the CODATA recommended values with the 1998^{73,74} and 2002⁷⁵ updates, followed by Mohr, Taylor, and Newell with the 2006^{76,77} and 2010^{2,3} updates. These updates are now

published every 4 year. The 1973, 1986, 1998, 2006, and 2010 CODATA updates were jointly published in *Reviews of Modern Physics* and the *Journal of Physical and Chemical Reference Data*, as will be the 2014 update (in preparation).^{78,79}

It is important to note the efforts by others who have worked both independently and in cooperation with the NRC and CODATA committees over the years to make recommendations and provide helpful discussions and debate over evaluation of the fundamental constants, including Gorbatesvich and coworkers (1967 and 1970),^{80,81} Levyleblond (1977),⁸² Pipkin and Ritter (1983),⁸³ Petley and coworkers (1989 and 2001),^{84,85} Delaeter (1994),⁸⁶ Uzan (2003),⁸⁷ Conroy (2003),⁸⁸ and Karshenboim (2005 and 2013).^{89,90}

4. References

- ¹See <http://indico.gsi.de/conferenceDisplay.py?confId=2742> for Fundamental Constants Meeting; accessed June 3, 2015.
- ²P. J. Mohr, B. N. Taylor, and D. B. Newell, "CODATA recommended values of the fundamental physical constants: 2010," *J. Phys. Chem. Ref. Data* **41**, 043109 (2012).
- ³P. J. Mohr, B. N. Taylor, and D. B. Newell, "CODATA recommended values of the fundamental physical constants: 2010," *Rev. Mod. Phys.* **84**, 1527 (2012).
- ⁴See <http://www.codata.org/committees-and-groups/fundamental-physical-constants> for CODATA Task Group on the Fundamental Constants; accessed June 11, 2015.
- ⁵S. G. Karshenboim, P. J. Mohr, and D. B. Newell, "Advances in determination of fundamental constants," *J. Phys. Chem. Ref. Data* **44**, 031101 (2015).
- ⁶See <http://www.bipm.org/en/CGPM/db/17/1> for Resolution 1 of the 17th Conférence Générale des Poids et Mesures (1983); accessed June 10, 2015.
- ⁷P. Giacomo, "News from the BIPM," *Metrologia* **20**, 25 (1984).
- ⁸See <http://www.bipm.org/en/CGPM/db/15/2/> for Resolution 2 of the 15th Conférence Générale des Poids et Mesures (1975); accessed June 10, 2015.
- ⁹J. Terrian, "News from the Bureau International des Poids et Mesures," *Metrologia* **11**, 179 (1975).
- ¹⁰F. D. Rossini, "Values of the fundamental constants for chemistry," *Pure Appl. Chem.* **9**, 453 (1964).
- ¹¹Research Concerning Metrology and Fundamental Constants, Report Subcommittee of the Committee on Fundamental Constants (National Academy Press, Washington, D.C., 1983).
- ¹²International Research Council, in *International Critical tables of Numerical Data, Physics, Chemistry and Technology* (McGraw-Hill, New York, 1926), Vol. 1, pp. 17–18.
- ¹³F. Henning and W. Jaeger, "The general physical constants," in *Handbuch der Physik. Bd. 2: Elementare Einheiten und ihre Messung*, edited by K. Scheel and H. Geiger (Springer, Berlin, 1926), pp. 487–517.
- ¹⁴R. T. Birge, "The most probable value of certain basic constants," *Science* **64**, 180 (1926).
- ¹⁵R. T. Birge, "Probable values of the general physical constants," *Rev. Mod. Phys.* **1**, 1 (1929).
- ¹⁶R. T. Birge, "Probable values of e, h, e/m, and alpha," *Phys. Rev.* **40**, 228 (1932).
- ¹⁷R. T. Birge, "A new table of values of the general physical constants (As of August, 1941)," *Rev. Mod. Phys.* **13**, 233 (1941).
- ¹⁸R. T. Birge, "The general physical constants as of August 1941 with details on the velocity of light only," *Rep. Prog. Phys.* **8**, 90 (1941).
- ¹⁹R. T. Birge, "The calculation of errors by the method of least squares," *Phys. Rev.* **40**, 207 (1932).
- ²⁰R. T. Birge, "On the values of fundamental atomic constants," *Phys. Rev.* **52**, 241 (1937).
- ²¹R. T. Birge, "The 1944 values of certain atomic constants with particular reference to the electronic charge," *Am. J. Phys.* **13**, 63 (1945).
- ²²R. T. Birge, "A survey of the systematic evaluation of the universal physical constants," *Nuovo Cimento, Suppl.* **6**, 39 (1957).
- ²³R. T. Birge, "The most probable value of the Planck constant h," *Phys. Rev.* **14**, 361 (1919).
- ²⁴R. T. Birge, "The electronic charge e," *Nature* **123**, 318 (1929).
- ²⁵R. T. Birge, "Probable values of e, h, e/m and alpha," *Phys. Rev.* **40**, 228 (1932).
- ²⁶R. T. Birge, "Probable values of e, h, e/m, and alpha,—An Addition," *Phys. Rev.* **40**, 319 (1932).
- ²⁷R. T. Birge, "The value of e/m," *Phys. Rev.* **42**, 736 (1932).
- ²⁸R. T. Birge, "The value of e/m," *Nature* **133**, 648 (1934).
- ²⁹R. T. Birge, "The velocity of light," *Nature* **134**, 771 (1934).
- ³⁰R. T. Birge, "The value of the electronic charge," *Phys. Rev.* **48**, 918 (1935).
- ³¹R. T. Birge, "The present status of the value of e/m," *Phys. Rev.* **54**, 972 (1938).
- ³²R. T. Birge, "The values of e, e/m, h/e and alpha," *Phys. Rev.* **58**, 658 (1940).
- ³³R. T. Birge, "The values of R and of e/m, from the spectra of H, D, and He+," *Phys. Rev.* **60**, 766 (1941).
- ³⁴W. N. Born, "The values and inter-relationships of c, e, h, M-p, m(o), G, and R," *Philos. Mag.* **10**, 994 (1930).
- ³⁵R. A. Millikan, "The most probable 1930 values of the electron and related constants," *Phys. Rev.* **35**, 1231 (1930).
- ³⁶R. A. Millikan, "The probability values for electrons and the associated constants for 1938," *Ann. Phys.* **32**, 34 (1938).
- ³⁷W. N. Bond, "Fundamental physical constants," *Philos. Mag.* **22**, 624 (1936).
- ³⁸S. von Friesen, "On the values of fundamental atomic constants," *Proc. R. Soc. A* **160**, 424 (1937).
- ³⁹F. G. Dunnington, "The atomic constants—A revaluation and an analysis of the discrepancy," *Rev. Mod. Phys.* **11**, 65 (1939).
- ⁴⁰H. T. Wensel, "International temperature scale and some related physical constants," *J. Res. Natl. Bur. Stand.* **22**, 375 (1939).
- ⁴¹U. Stille, "Wahrscheinlicher Fehler und Fehlergrenze in Anwendung auf das Problem der Atomkonstanten," *Z. Phys.* **120**, 703 (1943).
- ⁴²U. Stille, "Größen und Energiemaße in der Atomphysik," *Z. Phys.* **125**, 174 (1948).
- ⁴³J. W. M. Dumond, "Our present dilemma regarding the values of the natural constants e, m and h—A new graphical method of presentation," *Phys. Rev.* **56**, 153 (1939).
- ⁴⁴J. W. M. Dumond and E. R. Cohen, "Our knowledge of the atomic constants F, N, m, and h in 1947, and of other constants derivable therefrom," *Rev. Mod. Phys.* **20**, 82 (1948).
- ⁴⁵J. W. M. Dumond and E. R. Cohen, "Our knowledge of the atomic constants F, N, m, and h in 1947, and of other constants derivable therefrom," *Erratum Rev. Mod. Phys.* **21**, 651 (1949).
- ⁴⁶W. M. Dumond, "Recent changes and addition in the consistency diagram of the natural atomic constants," *Phys. Rev.* **75**, 1267 (1949).
- ⁴⁷J. W. M. Dumond, "The 1947 values of the atomic constants and the revision of the Faraday constant," *Phys. Rev.* **77**, 411 (1950).
- ⁴⁸J. W. M. Dumond and E. R. Cohen, "Least-squares adjusted values of the atomic constants as of December 1950," *Phys. Rev.* **82**, 555 (1951).
- ⁴⁹E. R. Cohen, "The Rydberg constant and the atomic mass of the electron," *Phys. Rev.* **88**, 353 (1952).
- ⁵⁰J. W. M. Dumond and E. R. Cohen, "Least-squares adjustment of the atomic constants, 1952," *Rev. Mod. Phys.* **25**, 691 (1953).
- ⁵¹J. W. M. Dumond and E. R. Cohen, "Variety of our sources of information on Avogadro's number and other constants," *Phys. Rev.* **94**, 1790 (1954).
- ⁵²E. R. Cohen, J. W. M. Dumond, T. W. Layton, and J. S. Rollett, "Analysis of variance of the 1952 data on the atomic constants and a new adjustment, 1955," *Rev. Mod. Phys.* **27**, 363 (1955).
- ⁵³E. R. Cohen, K. M. Crowe, and J. W. M. Dumond, *Fundamental Constants of Physics* (InterScience Publishers, New York, 1957).
- ⁵⁴J. W. M. Dumond, "Present sources of precise information on the universal physical constants," *Nuovo Cimento* **6**, 68 (1957).
- ⁵⁵E. R. Cohen and J. W. M. Dumond, "Changes in the 1955 atomic constants occasioned by revision of mu_e/mu_0," *Phys. Rev. Lett.* **1**, 291 (1958).
- ⁵⁶J. W. M. Dumond, "Status of knowledge of the fundamental constants of physics and chemistry as of January 1959," *Ann. Phys.* **7**, 365 (1959).
- ⁵⁷E. R. Cohen and J. W. M. Dumond, "Our knowledge of fundamental constants of physics and chemistry in 1965," *Rev. Mod. Phys.* **37**, 537 (1965).
- ⁵⁸E. R. Cohen, "A review of recent work in determination of fundamental physical constants," *Nuovo Cimento Suppl.* **4**, 839 (1966).
- ⁵⁹J. A. Bearden and H. M. Watts, "A re-evaluation of the fundamental atomic constants," *Phys. Rev.* **81**, 73 (1951).

- ⁶⁰J. A. Bearden and H. M. Watts, "A new formulation of the least-squares calculation of atomic constants," *Phys. Rev.* **81**, 160 (1951).
- ⁶¹J. A. Bearden, M. D. Earle, J. M. Minkowski, and J. S. Thomsen, "Present status of the fundamental atomic constants," *Phys. Rev.* **91**, 482 (1953).
- ⁶²J. A. Bearden, M. D. Earle, J. M. Minkowski, and J. S. Thomsen, "Present status of the atomic constants," *Phys. Rev.* **93**, 629 (1954).
- ⁶³J. A. Bearden and J. S. Thomsen, "A survey of atomic constants," *Nuovo Cimento, Suppl.* **5**, 267 (1957).
- ⁶⁴J. A. Bearden and J. S. Thomsen, "Résumé of Atomic Constants," *Am. J. Phys.* **27**, 569 (1959).
- ⁶⁵F. D. Rossini, F. T. Gucker, H. L. Johnston, L. Pauling, and G. W. Vinal, "Status of the values of the fundamental constants for physical chemistry as of July 1, 1951," *J. Am. Chem. Soc.* **74**, 2699 (1952).
- ⁶⁶B. N. Taylor, W. H. Parker, and D. N. Langenberg, "Determination of e/h, using macroscopic quantum phase coherence in superconductors: Implications for quantum electrodynamics and the fundamental physical constants," *Rev. Mod. Phys.* **41**, 375 (1969).
- ⁶⁷B. N. Taylor, W. H. Parker, D. N. Langenberg, and A. Denenstein, "On the use of the AC Josephson effect to maintain standards of electromotive force," *Metrologia* **3**, 13 (1967).
- ⁶⁸E. R. Cohen and B. N. Taylor, "The 1973 least-squares adjustment of the fundamental constants," *J. Phys. Chem. Ref. Data* **2**, 663 (1973).
- ⁶⁹E. R. Cohen and B. N. Taylor, "The 1986 adjustment of the fundamental physical constants," *Rev. Mod. Phys.* **59**, 1121 (1987).
- ⁷⁰E. R. Cohen and B. N. Taylor, "The 1986 CODATA recommended values of the fundamental physical constants," *J. Phys. Chem. Ref. Data* **17**, 1795 (1988).
- ⁷¹E. R. Cohen, "Changes in the fundamental constants—past and future," *IEEE Trans. Instrum. Meas.* **38**, 167 (1989).
- ⁷²B. N. Taylor and E. R. Cohen, "Recommended values of the fundamental physical constants—a status report," *J. Res. Natl. Inst. Stand. Technol.* **95**, 497 (1990).
- ⁷³P. J. Mohr and B. N. Taylor, "CODATA recommended values of the fundamental physical constants: 1998," *J. Phys. Chem. Ref. Data* **28**, 1713 (1999).
- ⁷⁴P. J. Mohr and B. N. Taylor, "CODATA recommended values of the fundamental physical constants: 1998," *Rev. Mod. Phys.* **72**, 351 (2000).
- ⁷⁵P. J. Mohr and B. N. Taylor, "CODATA recommended values of the fundamental physical constants: 2002," *Rev. Mod. Phys.* **77**, 1 (2005).
- ⁷⁶P. J. Mohr, B. N. Taylor, and D. B. Newell, "CODATA recommended values of the fundamental physical constants: 2006," *Rev. Mod. Phys.* **80**, 633 (2008).
- ⁷⁷P. J. Mohr, B. N. Taylor, and D. B. Newell, "CODATA recommended values of the fundamental physical constants: 2006," *J. Phys. Chem. Ref. Data* **37**, 1187 (2008).
- ⁷⁸P. J. Mohr, B. N. Taylor, and D. B. Newell, "CODATA recommended values of the fundamental physical constants: 2014," *Rev. Mod. Phys.* (unpublished).
- ⁷⁹P. J. Mohr, B. N. Taylor, and D. B. Newell, "CODATA recommended values of the fundamental physical constants: 2014," *J. Phys. Chem. Ref. Data* (unpublished).
- ⁸⁰S. V. Gorbatsevich and T. L. Zalutskaya, "Principal physical constants—methods of their determination and numerical values," *Meas. Tech. USSR* **1**, 5 (1967).
- ⁸¹S. V. Gorbatsevich, V. M. Holin, and V. N. Nosal, "Values of fundamental physical constants according to agreements in 1969," *Meas. Tech. USSR* **8**, 1124 (1970).
- ⁸²J. M. Levyleblond, "Conceptual nature of physical constants," *Riv. Nuovo Cimento* **7**, 187 (1977).
- ⁸³F. M. Pipkin and R. C. Ritter, "Precision measurements and fundamental constants," *Science* **219**, 913 (1983).
- ⁸⁴B. W. Petley, "The values of the fundamental physical constants," *Sci. Prog.* **73**, 231 (1989).
- ⁸⁵J. L. Flowers and B. W. Petley, "Progress in our knowledge of the fundamental constants of physics," *Rep. Prog. Phys.* **64**, 1191 (2001).
- ⁸⁶J. R. Delaeter, "Atomic weights and fundamental constants," *Interdiscip. Sci. Rev.* **19**, 121 (1994).
- ⁸⁷J. P. Uzan, "The fundamental constants and their variation: Observational and theoretical status," *Rev. Mod. Phys.* **75**, 403 (2003).
- ⁸⁸R. S. Conroy, "Frequency standards, metrology and fundamental constants," *Contemp. Phys.* **44**, 99 (2003).
- ⁸⁹S. G. Karshenboim, "Precision physics of simple atoms: QED tests, nuclear structure and fundamental constants," *Phys. Rep.* **422**, 1 (2005).
- ⁹⁰S. G. Karshenboim, "Recent progress in determination of fundamental constants and fundamental physics at low energies," *Ann. Phys.* **525**, 472 (2013).
- ⁹¹See supplementary material at <http://dx.doi.org/10.1063/1.4926441> for the program of the Workshop on the Determination of the Fundamental Constants held on February 1-6, 2015 in Elvillle, Germany.